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Unit 1 –Chemical Changes and Structure Key Area –Formulae and reaction quantities. 1. Hydrogen peroxide is a useful bleaching agent and is contained in many hair dyes. Over time, the hair dye becomes less effective as the hydrogen peroxide decomposes forming water and oxygen.

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= 50 kJ mol<sup>-1</sup> and H = -100 kJ mol<sup>-1</sup> B E A = 50 kJ mol<sup>-1</sup> and H = +100 kJ mol<sup>-1</sup> C E A = 100 kJ mol<sup>-1</sup> and H = +50 kJ mol<sup>-1</sup> D E A = 100 kJ mol<sup>-1</sup> and H = -50 kJ mol<sup>-1</sup>

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CfE Higher Chemistry Unit 1: Chemical Changes and Structure

In general: Average rate = change in variable change in time It is worth mentioning at this stage that the unit of concentration of a solution is measured as moles per litre. In previous courses this may have been shortened to mol/ or even as mol/dm<sup>3</sup>At this level the same unit will always be written as mol<sup>-1</sup>.

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Unit 1: Organic Chemistry - Ms. Sandhu's Classroom

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Topic 1 - Atomic Structure and the ... - A-Level Chemistry

o a = 1, b = 3, c = 2 and d = 4 o a = 1, b = 3, c = 2 and d = 2 o a = 1, b = 2, c = 2 and d = 2. National 5 Chemistry Revision Page 25 Unit 1 – Chemical Changes & Structures. 6 – The Mole. a) Gram Formula Mass (GFM) The gram formula mass (GFM) of a substance is known as the mass of one mole.

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